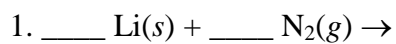
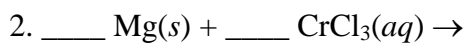


Types of Reactions – Ch. 8

For each of the following reactions, identify the reaction type, predict the products and balance the equation. Include physical states. Word equations must first be converted to formulas.



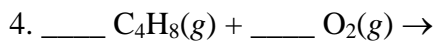
Reaction Type: _____



Reaction Type: _____



Reaction Type: _____



Reaction Type: _____

5. Aqueous solutions of potassium bromide and silver nitrate react to form a white precipitate.

Reaction Type: _____

6. Solid nickel is added to an aqueous solution of iron(II) sulfate.

Reaction Type: _____